

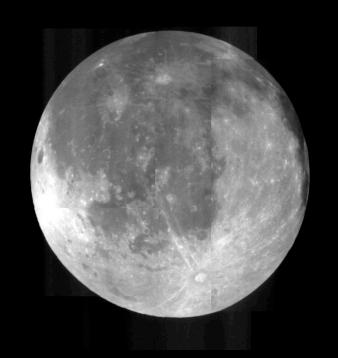
### **Dust on Mars**

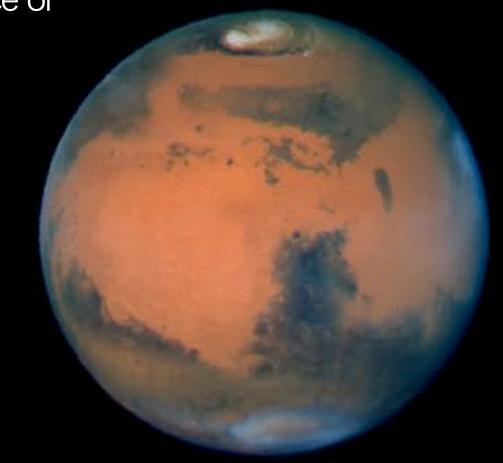
C. P. McKay
NASA Ames Research Center
cmckay@mail.arc.nasa.gov

#### Dust on the Moon and Mars?

- 1. What can we learn about Mars dust on the Moon?
- 2. Physical, chemical & biological properties of Mars dust
- 3. Future measurement needs

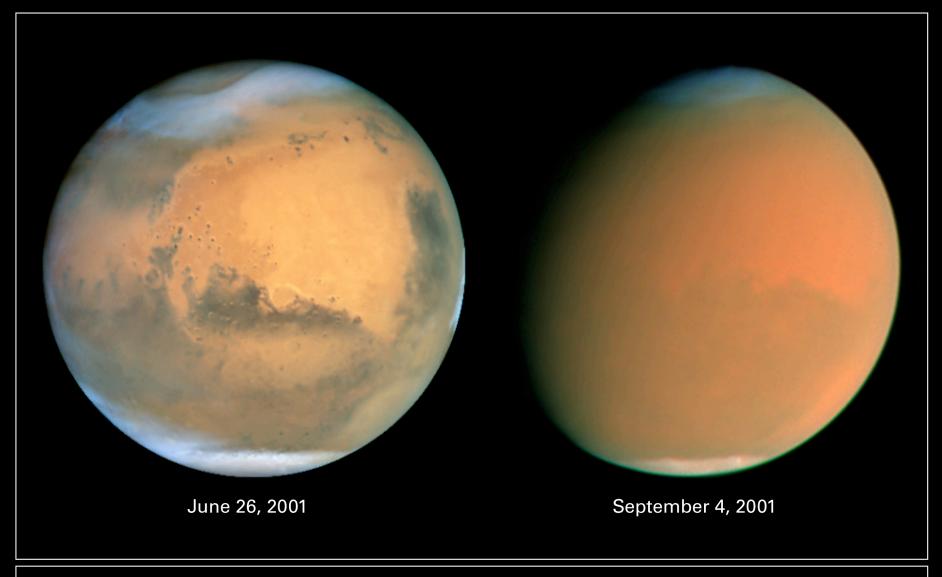
On the Moon and Mars the source of soil is comminuted basaltic rock



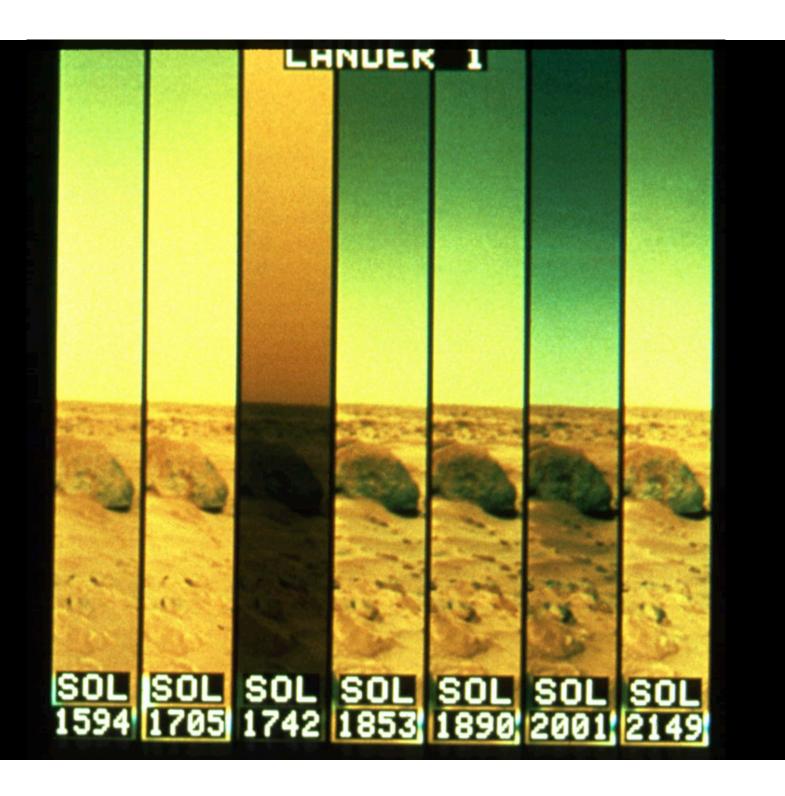


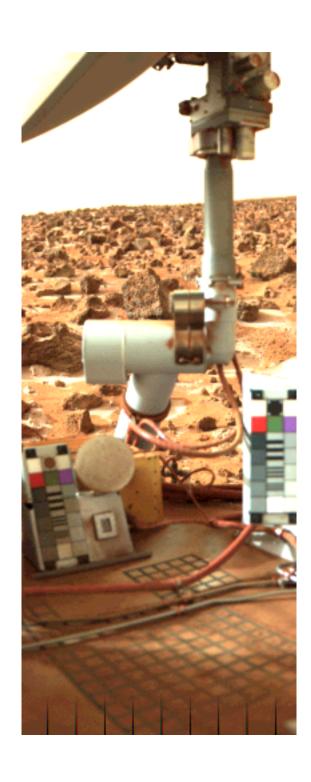
On Mars this has been further altered by:

- -Atmosphere & wind
- -lce, water, & brines
- -Oxidation
- -Volcanic gases
- -Life?



#### Mars • Global Dust Storm Hubble Space Telescope • WFPC2

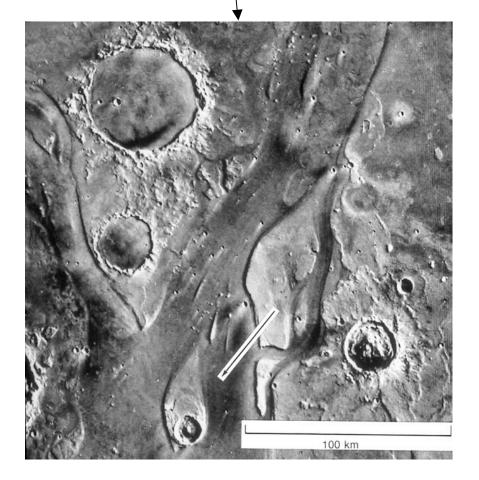


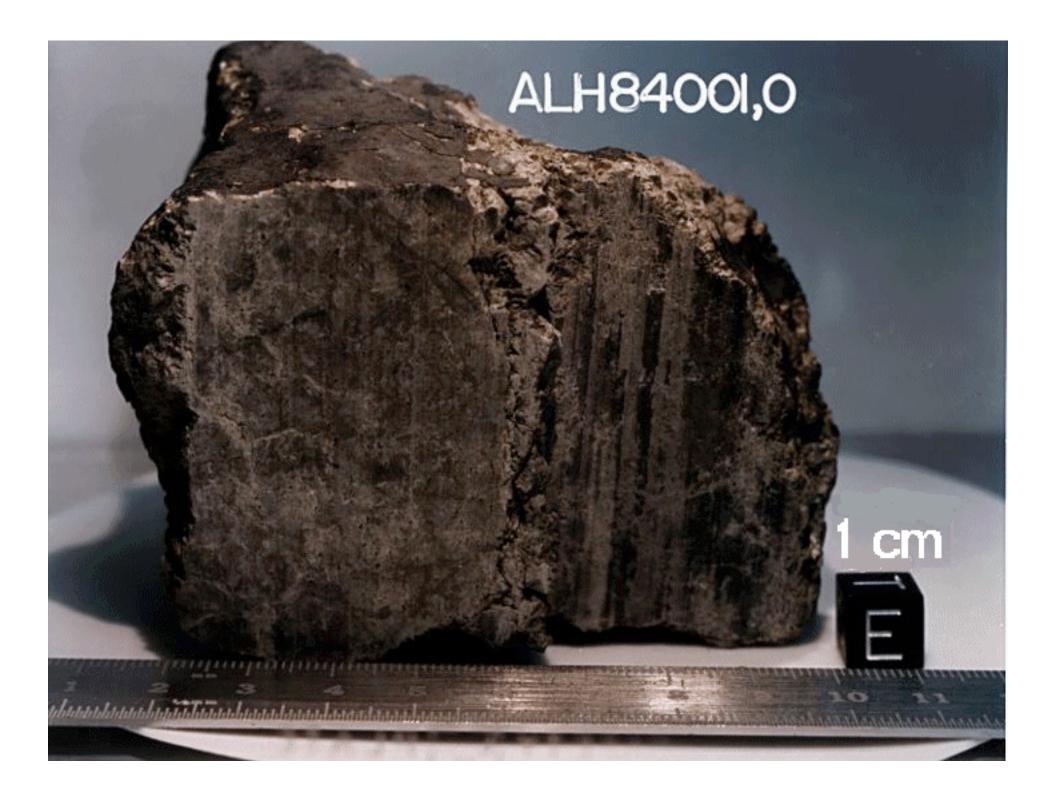


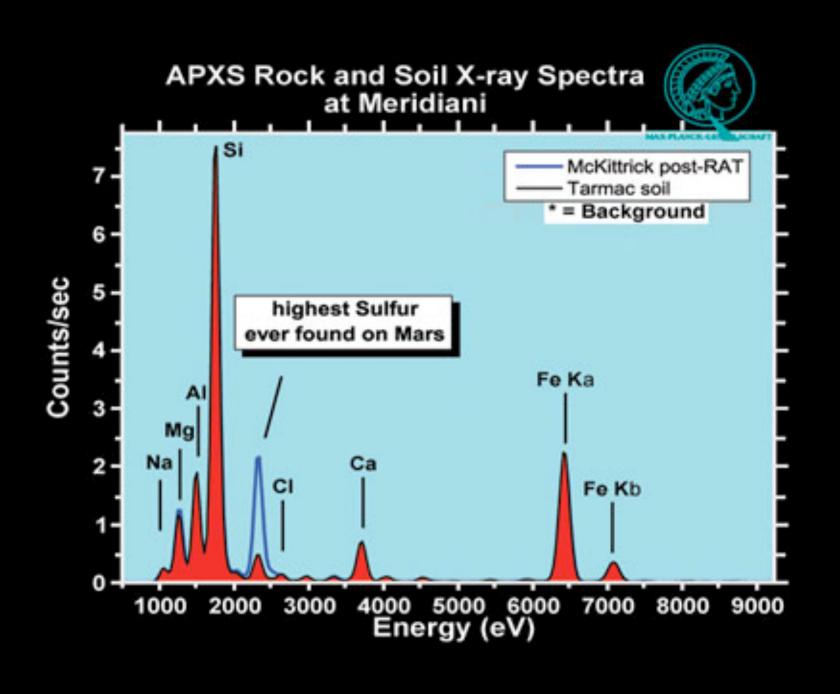
### Viking:

water frost on Mars

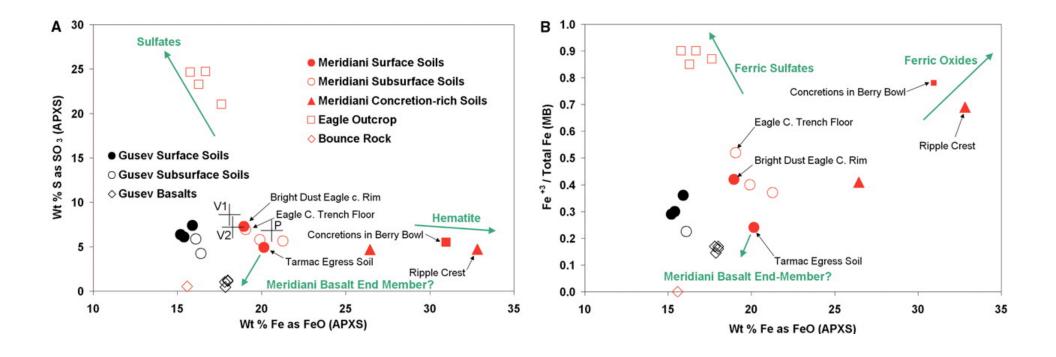
water flowed on the surface







## The soils at the two Viking lander sites, Pathfinder, and the two MER sites are similar in composition.

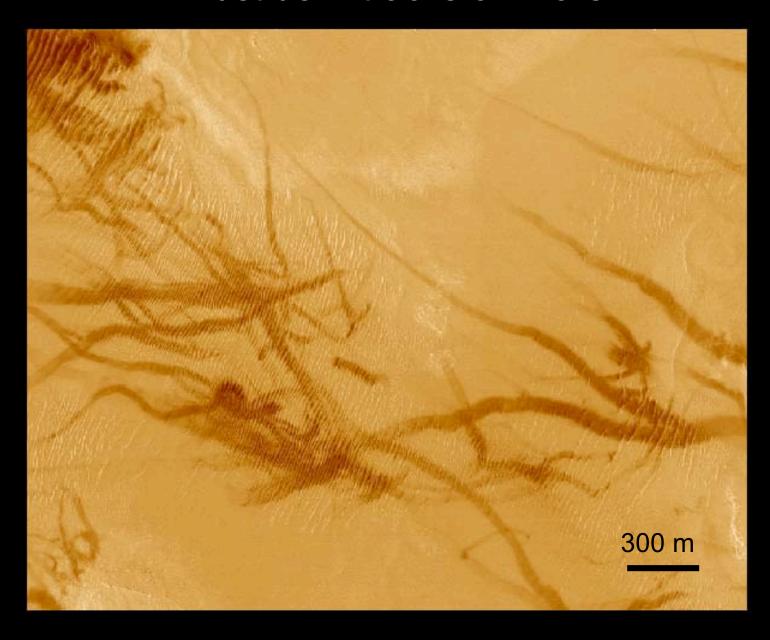


Soderblom et al. Science 2005

### Physical properties of Martian dust

- No direct data for sizes < 100 microns</li>
- Dust on Mars is likely to be physically weathered due to wind transport, i.e. shards or long thin slivers of particles are unlikely.
- Other than the presence of oxidants and the absence of organics, the martian soil is not unusual. No reason to expect toxic metals etc.

#### Dust devil tracks on Mars



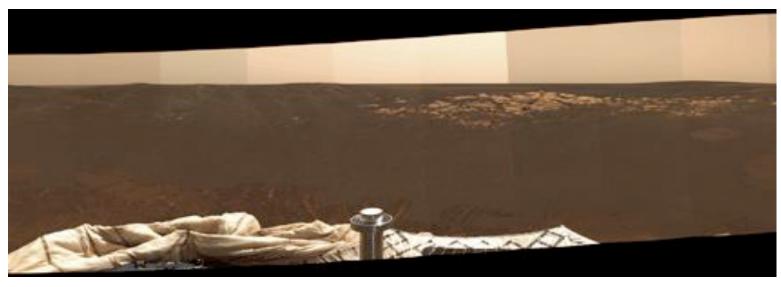




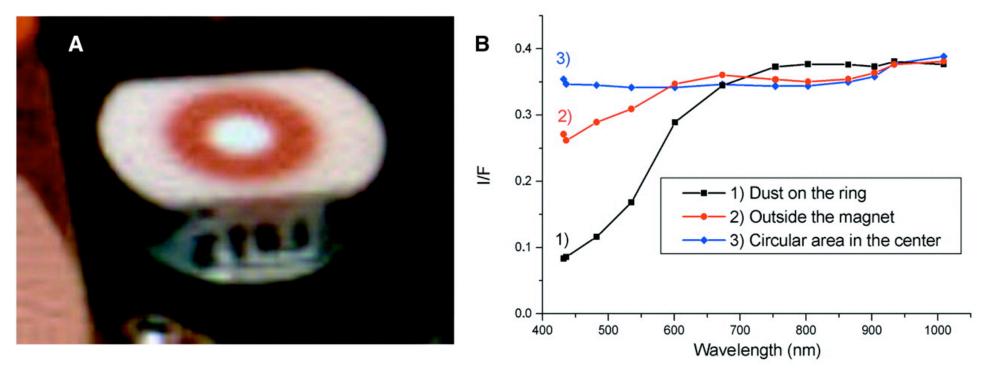
Before and After a "cleaning" event on Spirit March 2005



Clean airbags at Eagle Crater

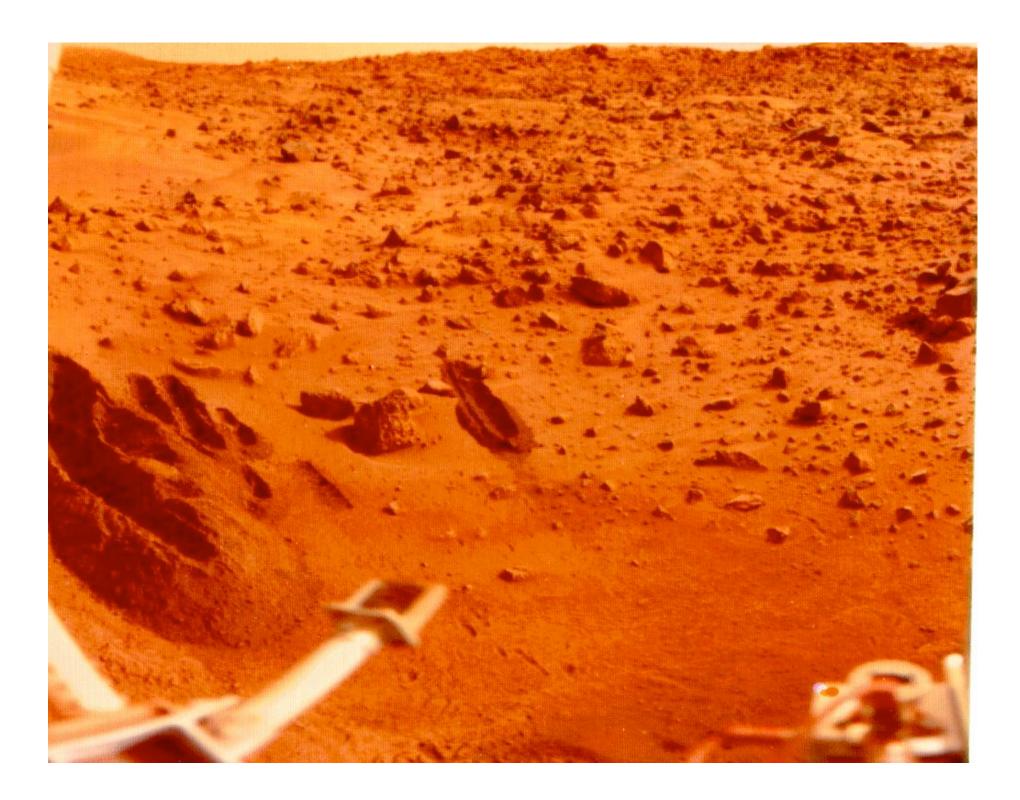


Virtually all martian airborne dust is magnetic and an area can be kept clear with magnetic fields.



The "sweep" magnet:

- 1. Ring of dust is magnetic particles
- 2. Outside the ring is all particles (no magnetic field)
- 3. Center area is non-magnetic particles



SOIL SAMPLE The Viking Biology Experiments DUPLICATE TO CONTROL LABELED CARBON DIOXIDE LABELED NUTRIENT 14CO2 WATER NUTRIENT MEDIUM He Kr MEDIUM/  $CO_2$ LIGHT SOURCE CHROMATOGRPAH SAMPLE SAMPLE SAMPLE COLUMN **PYROLYSIS** GAS DETECTOR DETECTOR DETECTOR FOR H<sub>2</sub> N<sub>2</sub> FOR C14 FOR C14 02 CH4 CO2 PYROLYTIC RELEASE Α LABELED RELEASE C GAS EXCHANGE

#### Viking results indicate soil reactivity

- O<sub>2</sub> released upon humidification (GEx)
- CO<sub>2</sub> released from organics (LR)
- No organics detected in the soil (GCMS)

### Oxidant levels indicated by Viking

Location	$GEx (KO_2 \rightarrow O_2)$	$LR(H_2O_2 \rightarrow O_2)$
Viking 1 (surface)	35 ppm/m	1ppm/m
Viking 2 (surface)	10	1
Viking 2 (sub-rock)	3	1

## Gas Exchange Experiment (GEx)

- 1. O<sub>2</sub> was released on humidification
- 2. Reactivity was unaltered by heating
- 3. Concentration varies: 3-35 ppm/m

Sample	GEx O <sub>2</sub>	Oxidant
	(nmoles cm <sup>-3</sup> )	$(KO_2 \rightarrow O_2)$
Viking 1 surface	770	35 ppm/m
Viking 2 surface	194	10
Viking 2 sub-rock	70	3

## Labeled Release (LR)

- 1. CO<sub>2</sub> was released from organics
- 2. Reactivity was effected by heating
  - -eliminated by 160°C for 3 hours
  - --eliminated by 50°C for 3 hours
  - -reduced 70% by 46°C for 3 hours
  - -eliminated by 141 days at 10°C
- 3. Release constant, possibly only formate consumed

Sample	LR CO <sub>2</sub>	Oxidant
	(nmoles cm <sup>-3</sup> )	$(H_2O_2 \rightarrow O)$
Viking 1 surface	~30	1 ppm/m
Viking 2 surface	~30	1
Viking 2 sub-rock	~30	1

## Suggested Explanation: Three different oxidants (Klein 1978, 1979)

- 1. A strong thermally stable oxidant eg. KO<sub>2</sub>, ZnO<sub>2</sub>, CaO<sub>2</sub> at 3-35 ppm/m
- 2. A strong thermally labile oxidant eg. H<sub>2</sub>O<sub>2</sub> at 1 ppm/m
- 3. A weak oxidant  $\gamma$ -Fe<sub>2</sub>O<sub>3</sub>

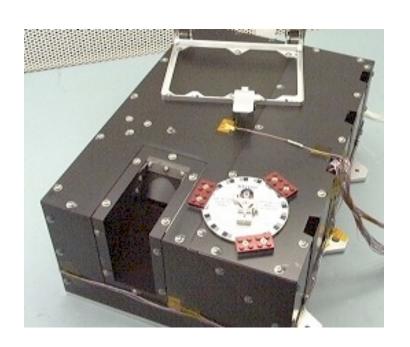
fundamental source: atmospheric H<sub>2</sub>O<sub>2</sub> produced by UV light, 2x10<sup>9</sup> molec cm<sup>-2</sup> s<sup>-1</sup>

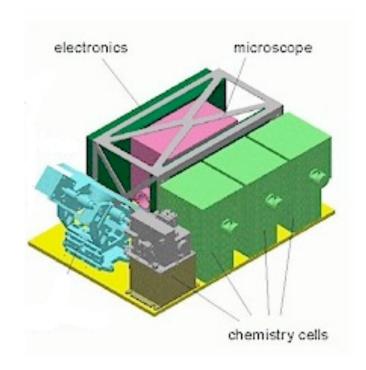
#### **Laboratory Simulations**

H<sub>2</sub>O<sub>2</sub> modified TiO<sub>2</sub> (Quinn and Zent, 1999) is probably the best laboratory simulation todate.

Yen et al. (2000) suggested superoxide ions

Microscopy, Electrochemistry, and Conductivity Analyzer (MECA) for the Phoenix mission (2007).



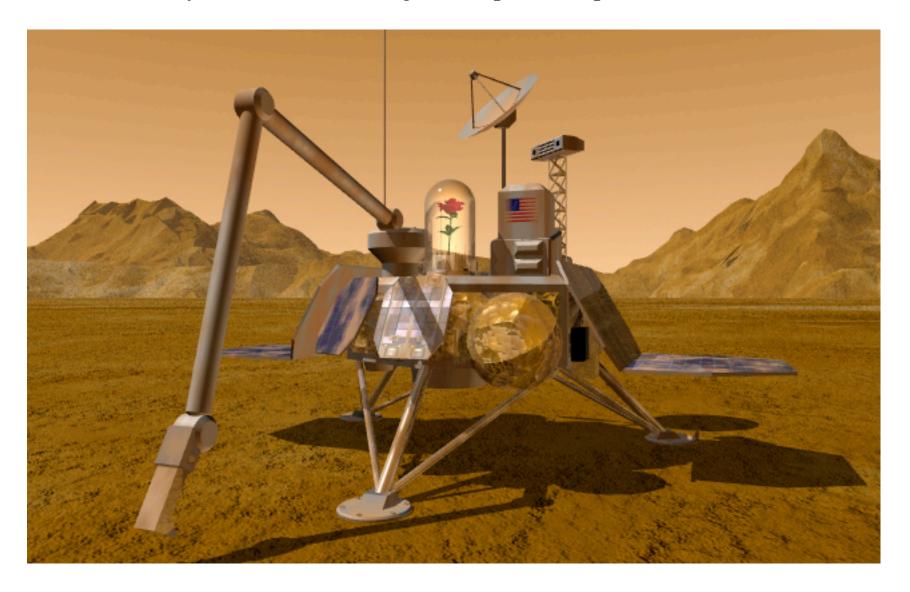


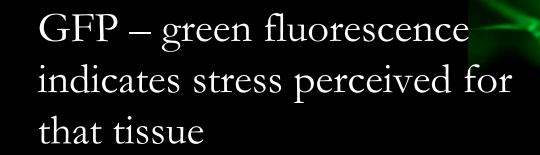
#### MECA comprises three main instruments:

- Wet Chemistry Laboratory consisting of four wet cells
- Optical and Atomic Force Microscopes
- Thermal and Electrical Conductivity Probe

#### Near-term missions

Soil toxicity based on whole organism (plant) response in the environment.







Plant expression Adh/GFP in response to hypobaric stress

Data from R. Ferl & A.L. Paul, Univ. Fla



#### Mars-like soils on Earth

Laboratory analog: H<sub>2</sub>O<sub>2</sub> modified TiO<sub>2</sub>

Geological analog: Palagonite from volcanoes in Hawaii & Iceland

Biological & organic analog: Atacama desert soil

#### Palagonite in Iceland





View of Northern Chile (NASA Space Shuttle)

View of Atacama From Shuttle with Hubble Telescope In the Foreground

-Yungay

Antofagasta



# Mars-like soils in the Atacama Desert, Chile

- Organics drop to low levels in the arid core.
- Oxidized organics increase in relative concentration
- No culturable microorganisms
- No recoverable DNA
- Abiotic consumption of organic material

